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Synthesis of OligoPhenyleneVinylene Dendrimers with Resorcinarene Core and their Supramolecular Complexes with Fullerene C₆₀

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Dendritic branches of Oligophenylvinylene chains have been attached to a resorcinarene core. The supramolecular complexes with fullerene C₆₀ were studied with two dendrimers of first and second generation. All the compounds were characterized by ¹H, ¹³C NMR, FTIR, UV–vis spectroscopy, MALDI-TOF, FAB + mass spectra and elemental analysis.

Keywords: OPV; π -Conjugated system; Resorcinarene; Dendrimers; Supramolecular complexes; Fullerene C₆₀

INTRODUCTION

Since the advent of dendrimers, a great variety of these well-defined structures have been synthesized [1] and now they represent the center of an important interdisciplinary field of research with possible technical applications in material science, catalysis, biotechnology, and medicine [2–4]. In recent years, there has been a considerable interest in incorporating π -conjugated systems into dendrimers aimed at an access to electroluminescent compounds [5,6]. Dendrimers with π -conjugated systems representing cores [7,8] or peripheral subunits [9] and included into each branching unit [10] have been documented. The core in the dendritic structures offers a unique opportunity to vary the physical and electronic properties independently, and the high molecular weight dendrimers possess a regulated nanospace. These well defined nanospaces can provide cavities for inclusion of big molecules. We are interested in resorcinarenes because of their ease of synthesis and the possibility to be obtained only in their *rccc*

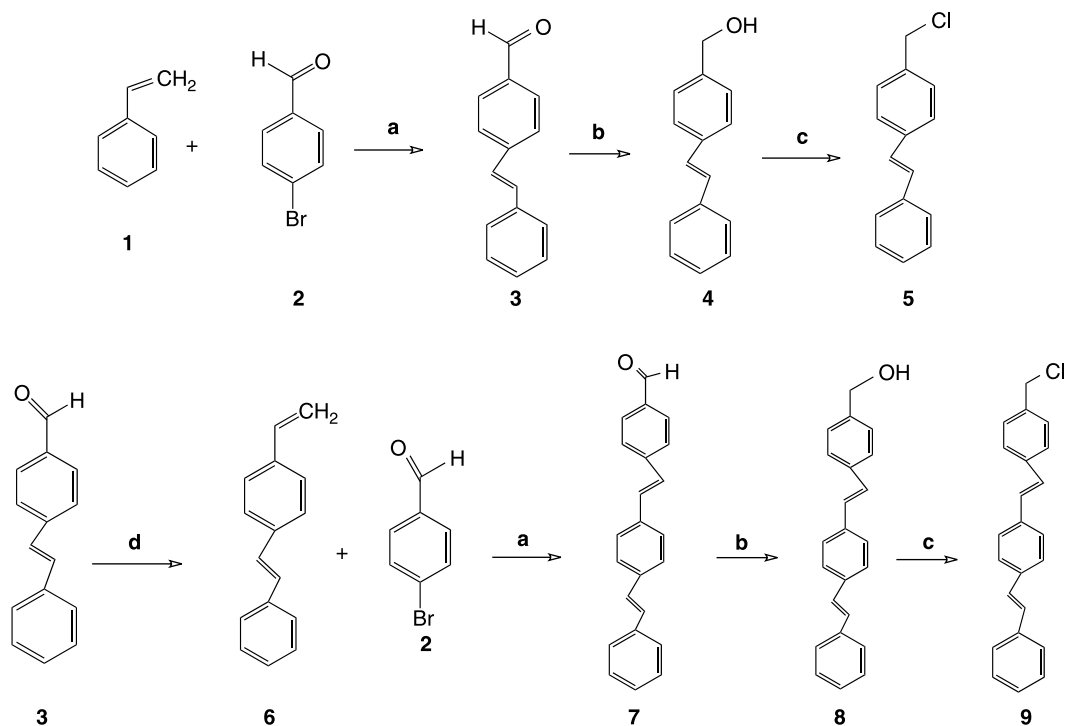
conformation, making them less affected by steric constraints [11]. In this paper we report the synthesis of dendrimers with π -conjugated systems oligophenylenevinylene (OPV) using resorcinarenes with an *rccc* conformation as core molecules and their supramolecular complexes with fullerene C₆₀.

RESULTS AND DISCUSSION

Dendrons containing phenyl vinyl groups were prepared according to the convergent Fréchet approach [12]. Styrene was obtained from Aldrich reagents and used in the Heck reaction coupling of the 4-bromo-benzaldehyde **2** in dimethyl formamide and triethylamine using palladium acetate as catalyst to afford **3**. The aldehyde **3** was reduced with LiAlH₄ in THF at 0°C to give alcohol **4**, which was converted into the chloride **5** upon treatment with thionyl chloride and pyridine in dichloromethane at 0°C. This chloride was used as the reagent for the synthesis of the first generation of stilbene containing dendrimers (Scheme 1).

After synthesizing the first generation, the higher generation can be formed by applying the same set of reactions; Wittig of **3** to obtain the (E)-1-styryl-4-vinylbenzene **6**, which reacted by Heck reaction with 4-bromo-benzaldehyde under the same conditions to afford aldehyde **7**, followed by the reduction of the aldehyde group to benzylic alcohol **8** and chlorination of the benzylic alcohol to obtain the dendron **9**. Dendrons **5** and **9** were characterized by ¹H- and ¹³C-NMR, IR, FAB + mass spectrometry. The ¹H

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SCHEME 1 Synthesis of lineal dendrons of first and second generation; a) $\text{Pd}(\text{OAc})_2/\text{TOP}$, DMF/ Et_3N , 120°C b) THF, LiAlH_4 , 0°C c) Py, CH_2Cl_2 , SOCl_2 , 0°C d) $\text{CH}_3(\text{Ph})_3\text{PBr}$, $n\text{-BuLi}$, THF, 10°C .

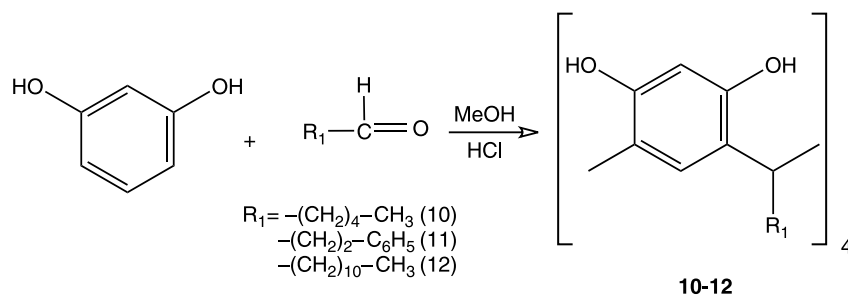
NMR spectra of the dendrons **5**, **9** contain signals at δ_{H} 4.60 for the methylene protons $\text{Ar}-\text{CH}_2-\text{Cl}$, for the vinylic protons were observed a singlet at δ_{H} 7.10. In agreement with the ^1H NMR data all the π -conjugated systems showed *trans*-configuration and the coupling constant was $J = 16.8$ Hz.

Resorcinarenes were obtained from resorcinol and three different aldehydes, hydrocinnamaldehyde, hexanal and dodecanal (Scheme 2). The structure of these resorcinarenes was confirmed by NMR and FAB+ mass spectrometry. Cyclic tetramers gave a well resolved triplet at δ_{H} 4.36, which is attributed to methine protons with a *rcc* conformation [11].

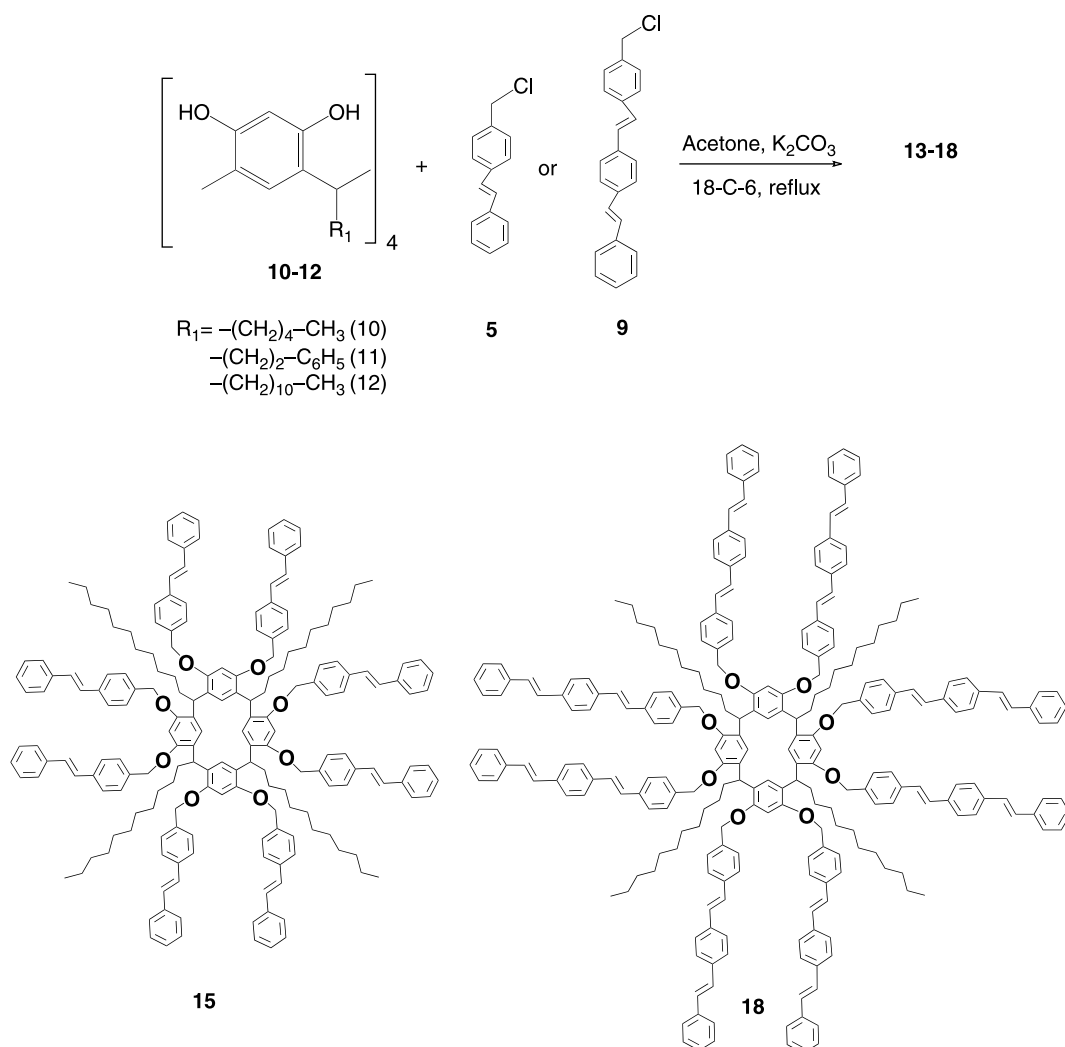
The iterative convergent strategy used for the styrene-focused dendrimers synthesis is shown in Scheme 3. As illustrated, the one step for dendrimer **13–18** formation involves a simple Williamson reaction between the dendrons **5** and **9** and resorcinarenes **10–12** in acetone and K_2CO_3 at reflux

for 3 days. The structures of **13–18** dendrimers were confirmed by ^1H - and ^{13}C -NMR, IR, and for the dendrimers **13**, **14** and **15** by MALDI-TOF mass spectrometry with all of the dendrimers having a molecular ion at the expected mass (Fig. 1). The molecular masses of the second generation dendrimers **16**, **17** and **18** could not be determined by MALDI-TOF mass spectrometry due to their poor volatility properties. However, gel-permeation chromatography (gpc) against polystyrene standards showed that all the dendrimers were monodisperse as would be expected from their well ordered construction.

The obtained **13–18** dendrimers are readily soluble in common organic solvents. The ^1H NMR spectrum of the dendrimer **15** showed one broad signal at δ_{H} 0.85 due to the CH_3 groups, two broad signals at δ_{H} 1.22 and 1.60 assigned to the CH_2 groups for the aliphatic chain, one singlet at δ_{H} 3.66

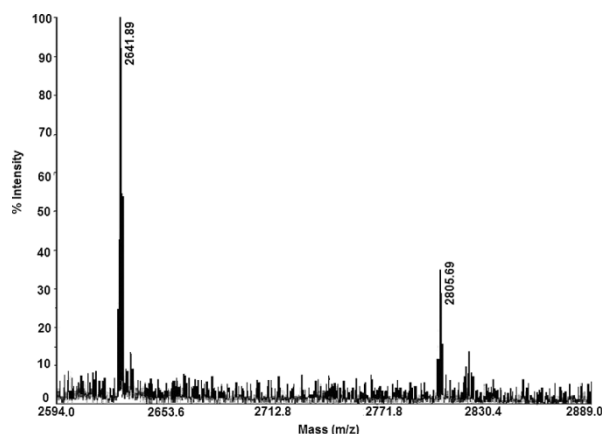


SCHEME 2 Synthesis of resorcinarenes.



SCHEME 3 Synthesis of dendrimers of first and second generation.

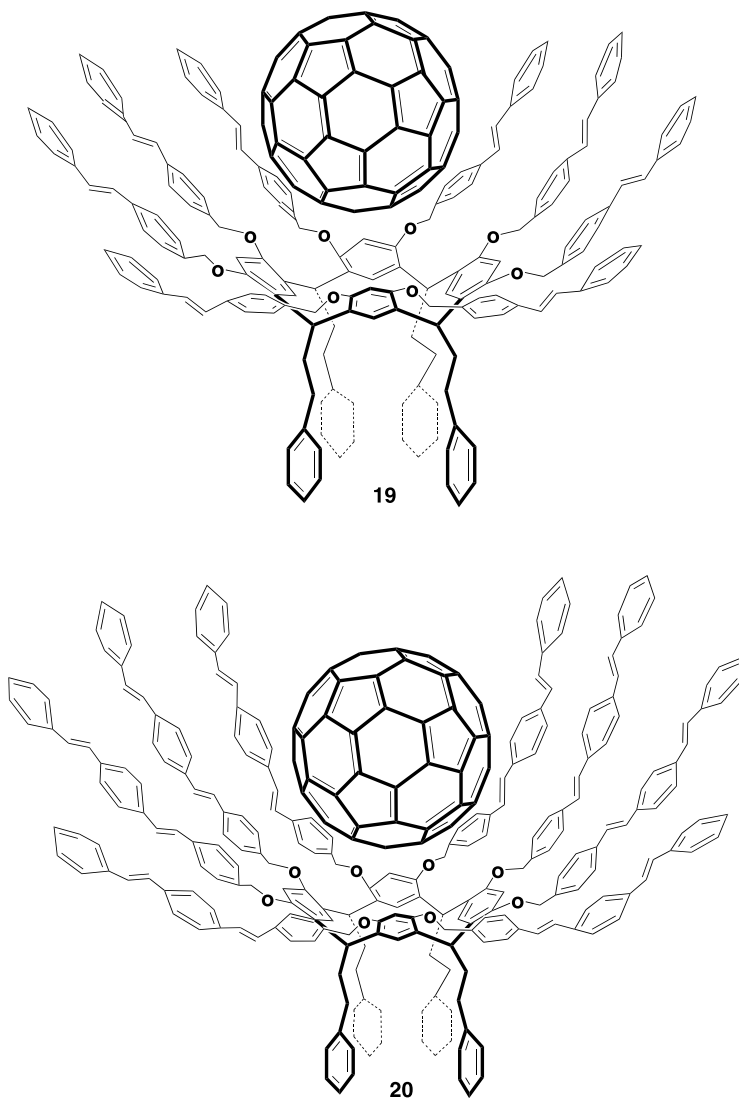
due to the methine protons at the resorcinarene ring, one broad signal at δ_H 5.13 for the $-CH_2-O$ protons, one multiplet at δ_H 7.01–7.15 due to the vinylic protons, and finally one broad signal at δ_H 7.26–7.48 for the aromatic protons was observed. The presence

FIGURE 1 MALDI-TOF mass spectrum of **15** first generation dendrimer.

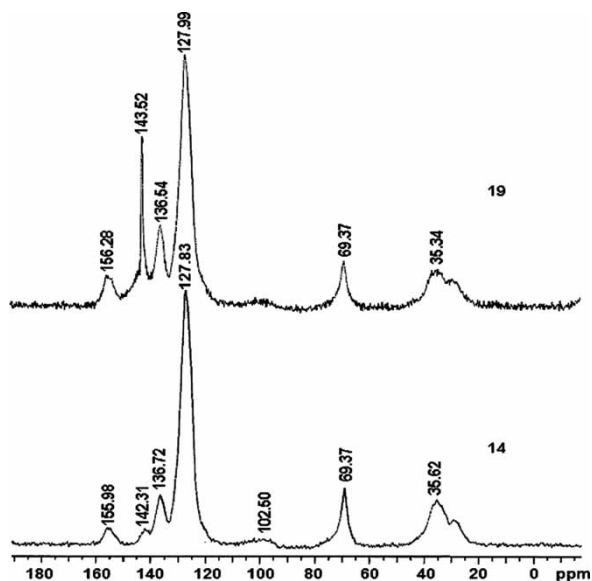
of one singlet at δ_H 3.66 confirmed the *rcc* conformation of the resorcinarene-dendrimers [11].

The supramolecular complexes were obtained from the resorcinarene-dendrimers **14** and **17** with fullerene C_{60} (Scheme 4). To a toluene solution of equimolar amounts of the fullerene, the dendrimer **14** or **17** was added. The reaction was stirred at reflux for 7 days. The solvent was evaporated in vacuum and the solid was carefully washed with small quantities of toluene to remove the traces of free fullerene and dendrimer, affording complexes **19** and **20** in 40 and 45% yields, respectively. Elemental analysis of the residual solid was consistent with the 1:1 stoichiometry.

The comparison of the solid-state ^{13}C CP-MAS NMR spectrum of **14** with that of the **14**: C_{60} 1:1 complex **19** shows some conformational changes in **14** (Fig. 2). The signal at δ_c 35.62 for the ethylene and methine groups was shifted upfield to δ_c 35.34. No changes for $Ar-CH_2-O$ groups were observed. The signal assigned for the carbons at the resorcinarene ring at δ_c 102.50 in the complex was diminished.



SCHEME 4 Supramolecular complexes 19 and 20.

FIGURE 2 ^{13}C CP-MAS NMR spectra of the resorcinarene-dendrimer 14 and of the supramolecular complex 19.

For the aromatic and vinylic carbons, the strong broad signal at δ_{c} 127.83 in the resorcinarene-dendrimer was shifted in the complex at δ_{c} 127.99. The signal observed at 136.72 in the compound 14 is shifted to upfield together with the fullerene signal at δ_{c} 136.54 in the complex 19, but the other signal at δ_{c} 142.31 for 14 is shifted downfield at δ_{c} 143.52 in 19. One additional signal for the $\text{C}_{\text{Ar}}\text{O}$ groups at δ_{c} 155.98 in the resorcinarene-dendrimer free was shifted at δ_{c} 156.28.

The solid state ^{13}C CP-MAS NMR spectrum of the 17: C_{60} 1:1 complex 20 shows some conformational changes for 17 (Fig. 3). For the ethylene and methine carbons the signal at δ_{c} 35.58 was shifted upfield at δ_{c} 33.64. The signal assigned to the $\text{Ar}-\text{CH}_2-\text{O}$ groups δ_{c} 69.61 was shifted 0.15 ppm. The signal assigned for the carbons at resorcinarene ring at δ_{c} 104.36 in the complex was diminished. For the aromatic carbons, two signals, one broad strong signal at δ_{c} 128.62 and one small signal at δ_{c} 135.55 due to the carbons C_{ipso} were observed. The signal assigned to the fullerene

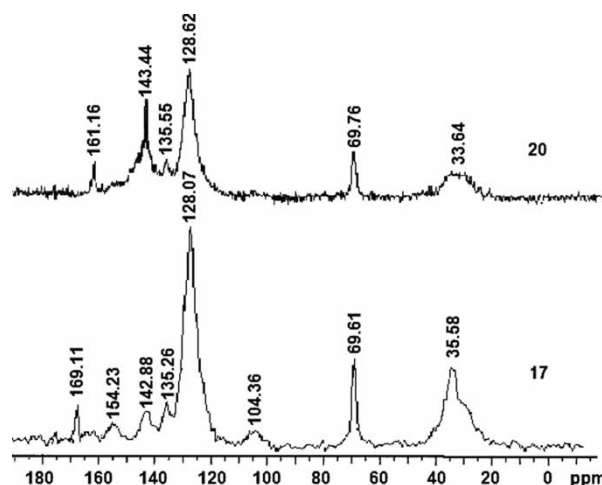


FIGURE 3 ^{13}C CP-MAS NMR spectra of the resorcinarene-dendrimer 17 and of the supramolecular complex 20.

was observed at δ_c 143.44, and finally the two signals for the $\text{C}_{\text{Ar}}\text{--O}$ groups at δ_c 154.23 and 169.11 observed in the resorcinarene-dendrimer free, in the complex appear together at δ_c 161.16.

CONCLUSIONS

Two generations of resorcinarene-dendrimers were synthesized by convergent building method obtaining relative good yields with preservation of the resorcinarene *rccc* conformation. The fixed nanospace constructed from the OPV system of 16 and 24 benzene rings around the resorcinarene core give the multipoint interaction to host the fullerene C_{60} . Two new supramolecular complexes between resorcinarene-dendrimers and fullerene C_{60} were also obtained, and $\pi\text{--}\pi$ $n\text{--}\pi$ interactions were observed in the complexes.

EXPERIMENTAL SECTION

Solvents and reagents were purchased as reagent grade and used without further purification. Acetone was distilled over calcium chloride. Tetrahydrofuran was distilled from sodium and benzophenone. Column chromatography was performed on Merck silica gel 60 Å (70–230 mesh). ^1H and ^{13}C NMR were recorded on a Varian-Unity-300 MHz with tetramethylsilane (TMS) as an internal reference. Infrared (IR) spectra were measured on a spectrophotometer Nicolet FT-SSX. Elemental analysis was determined by Galbraith Laboratories, INC Knoxville. FAB + mass spectra were taken on a JEOL JMS AX505 HA instrument. Matrix-assisted laser desorption/ionization were taken with a TofSpec spectrometer.

(E)-4-Styrylbenzaldehyde 3

A mixture of 1 2.8 g (27.02 mmol), 2 5 g (27.02 mmol), $\text{Pd}(\text{OAc})_2$ 3 mg (1.3 mmol), and tri-*o*-tolylphosphine POT 1 g (3.28 mmol) in $\text{Et}_3\text{N}/\text{DMF}$ 1:5 (120 mL) was stirred under N_2 at 120°C for 24 h. After cooling, the resulting mixture was filtered and the solvents evaporated. The crude product was purified by column chromatography (SiO_2 , hexane) to yield 5.4 g, (70%) as a yellow-pale powder, UV CH_2Cl_2 (nm): 231, 328. IR (KBr, cm^{-1}): 3431, 3024, 1624, 1445, 1119, 966, 828, 691. ^1H -NMR (CDCl_3), δ (ppm): 7.11 (d, 1H, CH= , $J = 16.2$ Hz), 7.25 (d, 1H, CH= , $J = 16.4$ Hz), 7.31–7.56 (m, 5H, Ar), 7.83 (d, 2H, Ar, $J = 1.8$ Hz), 7.87 (d, 2H, Ar, $J = 2.0$ Hz), 9.97 (s, 1H, --HC=O). ^{13}C NMR (CDCl_3), δ (ppm): 126.8 (Ar), 127.2 (CH=), 128.4 (CH=), 128.7 (Ar), 130.1 (Ar), 132.1 (Ar), 135.2 (Ar), 136.4 (Ar), 143.3 (Ar_{ipso}), 191.5 (C=O). Ms (m/z): 208. Anal. Calcd. for $\text{C}_{15}\text{H}_{12}\text{O}$: C 86.51, H 5.81%. Found: C 86.50, H 5.81%.

(E)-(4-Styrylphenyl)methanol 4

0.56 g (15.0 mmol) of 97% lithium aluminum hydride were dissolved in 50 mL of dry THF. To this emulsion, 2.1 g (10.0 mmol) of (E)-4-styrylbenzaldehyde 3 dissolved in 15 mL of dry THF were added drop wise using an addition funnel. The reaction was carried in acetone-ice bath for 4 h. After this time, 10 mL of water were added and the reaction mixture was filtered in Celite[®]. The solvent was evaporated and the residue was dissolved in dichloromethane. The resulting solution was dried with sodium sulfate, filtered and the product was vacuum dried to yield 1.5 g (70%) as a yellow powder, UV CH_2Cl_2 (nm): 232, 300, 313. IR (KBr, cm^{-1}): 3331, 3023, 2866, 1446, 1073, 997, 967, 780, 749, 690, 525. ^1H -NMR (CDCl_3), δ (ppm): 1.76 (s, 1H, --OH), 4.69 (s, 2H, CH_2), 7.10 (s, 2H, CH=), 7.25–7.40 (m, 5H, Ar), 7.49 (s, 2H, Ar), 7.53 (t, 2H, Ar, $J = 1.0$ Hz). ^{13}C NMR (CDCl_3), δ (ppm): 65.1 (CH_2), 126.5, 126.6, 127.3 (Ar), 127.6 (CH=), 128.2 (CH=), 128.6, 128.7 (Ar), 136.8, 137.2 (Ar), 140.2 (Ar_{ipso}). Ms (m/z): 210. Anal. Calcd. for $\text{C}_{15}\text{H}_{14}\text{O}$: C 85.68, H 6.71%. Found: C 85.67, H 6.70%.

(E)-1-(Chloromethyl)-4-styrylbenzene 5

3 g (14.0 mmol) of (E)-(4-styrylphenyl) methanol 4, 1 mL (14.0 mmol) of pyridine and 1.45 mL (14.0 mmol) of SOCl_2 were dissolved in 100 mL of dry CH_2Cl_2 , and this mixture was cooled to 10°C . The reaction was carried out under nitrogen in ice bath for 7 h. After this period, the solvent was evaporated and the resulting oil was dry supported and purified in a silica gel (60–240 pore size) column using a mixture of hexane- CH_2Cl_2 2:1 as eluent to yield 2.9 g (95%) as a yellow-brown powder, UV CHCl_3 (nm): 241, 304, 315. IR (KBr, cm^{-1}): 3025, 2958, 1489, 1446, 1264, 967, 821, 755, 691, 667, 539. ^1H -NMR (CDCl_3), δ (ppm):

4.60 (s, 2H, CH₂), 7.10 (s, 2H, CH=), 7.26–7.39 (m, 5H, Ar), 7.45–7.54 (m, 4H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 46.0 (CH₂), 126.5, 126.8 (Ar), 127.8 (CH=), 128.0 (Ar), 128.7 (Ar), 129.5 (Ar), 130.1 (Ar_{ipso}). Ms (m/z): 228 m/z. Anal. Calcd. for C₁₅H₁₃O: C 78.77, H 5.73%. Found: C 78.76, H 5.71%.

(E)-1-Styryl-4-vinylbenzene 6

n-BuLi 1.6 g (25.0 mmol) was added to a solution of **3** 5.2 g (25.00 mmol) and methyltriphenylphosphonium bromide 8.94 g (25.00 mmol) in dry THF (100 mL) at –10°C. The solution was stirred for 12 h, then a few drops of water were added and the resulting mixture was concentrated. The organic layer was extracted with CH₂Cl₂ and dried with Na₂SO₄, evaporated and purified by column chromatography (SiO₂, CH₂Cl₂/hexane 2:8) to give the (E)-1-styryl-4-vinylbenzene **6**, in 95% (5.0 g) yield as a yellow powder, UV CHCl₃ (nm): 242, 328. IR (KBr, cm⁻¹): 3080, 3023, 2924, 1914, 1815, 1624, 1448, 1405, 1073, 993, 965, 904, 521. ¹H-NMR (CDCl₃), δ (ppm): 5.25 (d, 1H, CH₂=, *J* = 11.0 Hz), 5.80 (d, 1H, CH₂=, *J* = 17.6 Hz), 6.71 (q, 1H, CH=), 7.10 (s, 2H, CH=), 7.26–7.42 (m, 5H, Ar), 7.46–7.54 (m, 4H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 113.7 (CH₂=), 126.4, 126.5, 126.6, 127.6 (Ar), 128.2 (CH=), 128.6 (Ar), 136.4 (Ar), 136.8 (Ar), 137.3 (Ar_{ipso}). Ms (m/z): 206 m/z. Anal. Calcd. for C₁₆H₁₄: C 93.16, H 6.84%. Found: C 93.14, H 6.84%.

4-((E)-4-(E)-Styrylstyryl)benzaldehyde 7

A mixture of **6** 2 g (9.7 mmol), **2** 1.8 g (9.7 mmol), Pd(OAc)₂ 3 mg (1.3 mmol), and tri-*o*-tolylphosphine POT 0.9 g (2.6 mmol) in Et₃N/DMF 1:5 (60 mL) was stirred under N₂ at 120°C for 24 h. After cooling, the resulting mixture was filtered and the solvents evaporated. The crude product was purified by column chromatography (SiO₂, hexane) to yield 2.3 g (60%) of 4-((E)-4-(E)-styrylstyryl)benzaldehyde **7** as a yellow powder, UV CHCl₃ (nm): 245, 376. IR (KBr, cm⁻¹): 3023, 2825, 1698, 1595, 1166, 967, 826, 545. ¹H NMR (CDCl₃), δ (ppm): 7.16 (d, 2H, CH=, *J* = 14.7 Hz), 7.18 (d, 2H, CH=, *J* = 16.5 Hz), 7.24–7.27 (m, 2H, Ar), 7.30 (t, 1H, Ar, *J* = 1.0 Hz), 7.35–7.40 (m, 2H, Ar), 7.52–7.57 (m, 4H, Ar), 7.66 (d, 2H, Ar, *J* = 6 Hz), 7.87 (d, 2H, Ar, *J* = 9 Hz), 10.00 (s, 1H, C=O). ¹³C NMR (CDCl₃), δ (ppm): 126.5, 126.8, 126.9, 127.1 (Ar), 127.2 (CH=), 127.8 (CH=), 128.7, 129.2, 130.2, 131.7, 135.3, 135.8, 137.2, 137.6, 143.4 (Ar). Ms (m/z): 310 m/z. Anal. Calcd. for C₂₃H₁₈O: C 89.00, H 5.85%. Found: C 89.01, H 5.85%.

4-((E)-4-(E)-Styrylstyryl)phenyl) methanol 8

0.56 g (15.0 mmol) of 97% lithium aluminum hydride were dissolved in 50 mL of dry THF. To this emulsion, 3 g (10.0 mmol) of 4-((E)-4-(E)-styrylstyryl)benzaldehyde **7** dissolved in 15 mL of dry THF were

added drop wise using an addition funnel. The reaction was carried out in acetone-ice bath for 4 h. After this time, 10 mL of water were added and the reaction mixture was filtered in Celite®. The solvent was evaporated and the residue was dissolved in dichloromethane. The resulting solution was dried with sodium sulfate, filtered and the product was vacuum dried to yield 3.6 g (70%) as a yellow powder, UV CHCl₃ (nm): 242, 358. IR (KBr, cm⁻¹): 3326, 2959, 2872, 1728, 1285, 1073, 966, 748, 548. ¹H NMR (CDCl₃), δ (ppm): 4.69 (s, 2H, CH₂), 4.74 (s, 1H, –OH), 7.12 (s, 1H, Ar), 7.29 (br, 4H, CH=), 7.33 (br, 2H, Ar), 7.36 (br, 2H, Ar), 7.39 (br, 2H, Ar), 7.51 (br, 4H, Ar), 7.54 (br, 2H, Ar). ¹³C NMR (CDCl₃, DMSO-*d*₆), δ (ppm): 64.7 (CH₂–OH), 109.2 (Ar), 126.5, 126.7, 128.1, 128.6 (Ar). Ms (m/z): 312 m/z. Anal. Calcd. for C₂₃H₂₀O: C 88.43, H 6.45%. Found: C 88.43, H 6.44%.

1-(Chloromethyl)-4-((E)-4-(E)-styrylstyryl)benzene 9

4.3 g (14.0 mmol) of 4-((E)-4-(E)-styrylstyryl)phenyl) methanol **8**, 1.12 mL (14.0 mmol) of pyridine and 1 mL (14.0 mmol) of SOCl₂ were dissolved in 100 mL of dry CH₂Cl₂, this mixture was cooled to 10°C. The reaction was carried out under nitrogen in ice bath for 7 h. After this period, the solvent was evaporated and the resulting oil was dry supported and purified in a silica gel (60–240 pore size) column using a mixture of hexane–dichloromethane 2:1 as eluent to yield 4.1 g (95%) as a yellow powder, UV CHCl₃ (nm): 243, 358. IR (KBr, cm⁻¹): 3361, 3023, 2945, 1635, 1513, 1420, 1087, 966, 825. ¹H NMR (DMSO-*d*₆), δ (ppm): 4.49 (s, 2H, CH₂), 7.26 (br, 1H, Ar), 7.30 (d, 2H, CH=, *J* = 16.8 Hz), 7.35 (d, 2H, CH=, *J* = 16.8 Hz), 7.41 (br, 2H, Ar), 7.54 (br, 4H, Ar), 7.58–7.62 (br, 6H, Ar). ¹³C NMR (DMSO-*d*₆), δ (ppm): 46.0 (CH₂), 126.4 (CH=), 126.6 (Ar), 127.7 (Ar), 128.6 (Ar), 128.9 (Ar), 129.2 (Ar), 136.8 (Ar), 137.0 (Ar). Ms (m/z): 330 m/z. Anal. Calcd. for C₂₃H₁₉Cl: C 83.50, H 5.79%. Found: C 83.52, H 5.79%.

The resorcinarenes were obtained in agreement with reference [11].

General Procedure for Dendrimers

A mixture of 1 mmol of the respective monochloride generation of dendron **5** or **9**, potassium carbonate (21.2 mmol) and 18-crown-6 0.56 g (2.12 mmol) in dry acetone (80 mL) was heated to reflux and stirred vigorously under nitrogen after 20 min. The compounds **10–12** (0.0125 mmol) dissolved in dry acetone (40 mL) were added dropwise and the reaction was continued for 7 days. The mixture was allowed to cool and the precipitate was filtered. The filtrate was evaporated to dryness under reduced pressure. The residue dissolved in diethyl ether was washed with an aqueous solution of 5% Na₂CO₃ (3 times). The organic layer was dried

and evaporated to dryness and reprecipitated with dichloromethane–methanol.

Dendrimer G1 13. Resorcinarene **10**; **5**; brown powder 0.29 g (90%), UV CHCl₃ (nm): 242, 302, 315. IR (KBr, cm⁻¹): 3416, 3054, 2926, 2858, 1606, 1494, 1290, 1117, 962. ¹H NMR (CDCl₃), δ (ppm): 0.83 (br, 12H, CH₃), 1.27 (br, 24H, CH₂), 2.17 (br, 8H, CH₂), 3.60 (br, 4H, CH), 4.82 (br, 16H, CH₂–O), 6.42 (br, 4H, Ar), 7.12 (br, 16H, CH=), 7.25 (br, 76H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 14.1 (CH₃), 22.7 (CH₂), 27.7 (CH₂), 32.2 (CH), 70.0 (CH₂–O), 101.0 (Ar), 126.4 (Ar), 127.5 (CH=), 128.6 (Ar), 131.0 (Ar ipso), 134.0 (Ar_{ipso}), 136.4 (Ar), 137.2 (Ar), 139.3 (Ar_{ipso}), 154.3 (Ar–O). MALDI TOF: 2305 m/z. Anal. Calcd. for C₁₆₈H₁₆₀O₈: C 87.46, H 6.99%. Found: C 87.45, H 6.98%.

Dendrimer G1 14. Resorcinarene **11**; **5**; brown–red powder 0.21 g (91%), UV CHCl₃ (nm): 242, 301, 315. IR (KBr, cm⁻¹): 3427, 3025, 2918, 1607, 1495, 1291, 1107. ¹H RMN (CDCl₃), δ (ppm): 2.31 (br, 8H, CH₂), 2.78 (br, 8H, CH₂) 3.62 (s, 16H, CH₂–O), 4.82–4.91 (t, 4H, CH, *J* = 1.5 Hz), 6.45 (s, 4H, Ar), 7.01 (br, 16H, CH=), 7.06–7.43 (m, 96H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 34.6 (CH₂), 34.9 (CH), 70.0 (CH₂–O), 126.5 (CH=), 127.5 (Ar), 128.3 (Ar), 128.6 (Ar), 136.5 (Ar), 137.1 (Ar). ¹³C CP-MAS NMR δ (ppm): 35.62 (CH), 69.37 (CH₂–O), 102.50 (Ar), 127.83 (CH=, Ar), 136.72 (Ar), 142.31 (Ar), 155.98 (Ar–O). MALDI TOF: 2441 m/z. Anal. Calcd. for C₁₈₀H₁₅₂O₈: C 88.49, H 6.27%. Found: C 88.50, H 6.27%.

Dendrimer G1 15. Resorcinarene **12**; **5** brown powder 0.20 g (85%), UV CHCl₃ (nm): 242, 301, 315. IR (KBr, cm⁻¹): 3419, 2924, 2852, 1497, 1108, 961. ¹H NMR (CDCl₃), δ (ppm): 0.85 (br, 12H, CH₃), 1.22 (br, 72H, CH₂), 1.60 (br, 8H, CH₂), 3.66 (s, 4H, CH), 5.13 (br, 16H, CH₂–O), 7.01–7.15 (m, 16H, CH=), 7.26–7.48 (br, 80H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 14.1 (CH₃), 22.6 (CH₂), 29.4 (CH₂), 29.7 (CH₂), 31.9 (CH), 70.2 (CH₂–O), 126.4 (CH=), 127.5 (Ar), 128.6 (Ar), 137.1 (Ar). MALDI TOF: 2641 m/z. Anal. Calcd. for C₁₉₂H₂₀₈O₈: C 87.23, H 7.93%. Found: C 88.25, H 7.92%.

Dendrimer G2 16. Resorcinarene **10**; **9** brown powder 0.38 g (90%), UV CHCl₃ (nm): 243, 296, 359. IR (KBr, cm⁻¹): 3422, 2925, 1613, 1461, 1107, 962. ¹H NMR (CDCl₃), δ (ppm): 0.89 (br, 12H, CH₃), 1.25 (br, 24H, CH₂), 2.19 (br, 8H, CH₂), 3.59 (br, 16H, CH₂–O), 4.33 (t, 4H, CH, *J* = 1.5 Hz), 6.13 (br, 4H, Ar), 7.12 (br, 32H, CH=), 7.26–7.38 (m, 76H, Ar), 7.52 (br, 32H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 13.5 (CH₃), 22.2 (CH₂), 27.6 (CH₂), 28.7 (CH₂), 31.6 (CH), 69.4 (CH₂–O), 125.9 (CH=), 126.3 (Ar), 126.7 (Ar), 127.1 (Ar), 127.6 (Ar), 128.2 (Ar), 136.0 (Ar). Anal. Calcd. for C₂₃₂H₂₀₈O₈: C 89.19, H 6.71%. Found: C 89.19, H 6.72%.

Dendrimer G2 17. Resorcinarene **11**; **9** brown–red powder 0.31 g (92%), UV CHCl₃ (nm): 243, 297, 359. IR (KBr, cm⁻¹): 3386, 2923, 1654, 1400, 1106, 961. ¹H NMR (CDCl₃), δ (ppm): 2.63 (br, 16H, CH₂), 3.51 (br, 16H, CH₂–O), 4.56 (br, 4H, CH), 6.32 (br, 4H, Ar), 6.92

(br, 32H, CH=), 7.12–7.54 (m, 128H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 13.5 (CH₃), 22.2 (CH₂), 31.2 (CH₂), 31.6 (CH), 69.9 (CH₂–O), 122.0 (Ar), 125.6 (Ar), 126.5 (Ar), 126.8 (Ar), 128.3 (CH=), 130.8 (Ar), 136.6 (Ar), 137.3 (Ar), 142.9 (Ar), 152.6 (Ar). ¹³C CP-MAS NMR δ (ppm): 35.58 (CH), 69.61 (CH₂–O), 104.36 (Ar), 128.07 (CH=, Ar), 135.26 (Ar), 142.88 (Ar), 154.23 (Ar–O), 169.11 (Ar–O). Anal. Calcd. for C₂₄₄H₂₀₀O₈: C 89.89, H 6.18%. Found: C 89.89, H 6.16%.

Dendrimer G2 18. Resorcinarene **12**; **9** brown powder 0.30 g (90%), UV CHCl₃ (nm): 243, 359. IR (KBr, cm⁻¹): 3396, 2924, 2853, 1623, 1458, 1253, 1108, 963. ¹H NMR (CDCl₃), δ (ppm): 0.85 (br, 12H, CH₃), 1.25 (br, 72H, CH₂), 2.11 (br, 8H, CH₂), 3.60 (br, 16H, CH₂–O), 4.33 (m, 4H, CH), 6.13 (br, 4H, Ar), 7.12 (br, 32H, CH=), 7.26–7.52 (m, 108H, Ar). ¹³C NMR (CDCl₃), δ (ppm): 13.5 (CH₃), 22.1 (CH₂), 29.5 (CH₂), 69.4 (CH₂–O), 125.9 (Ar), 126.3 (Ar), 128.2 (Ar), 135.6 (Ar), 136.8 (Ar). Anal. Calcd. for C₂₅₆H₂₅₆O₈: C 88.85, H 7.46%. Found: C 89.87, H 7.46%.

Complexes 19 and 20

A solution of **14** or **17** (0.138 mmol) in toluene (50 ml) was added to a toluene solution (100 ml) of C₆₀ (0.138 mmol), and the mixture was stirred vigorously at 80°C for 7 days. After this period, the solvent was evaporated to dryness and the solid obtained was carefully washed with small quantities of toluene.

Supramolecular complex 19

Brown powder 80 mg (40% yield). UV–Vis (solid state) nm: 249, 365. IR (cm⁻¹), 3430, 2927, 3055, 3024, 2916, 2862, 1602, 1494, 1426, 1288, 1178, 1104, 1014, 959, 807, 747, 694, 574, 524. ¹³C CP-MAS NMR δ_c (ppm): 35.34 (CH, CH₂), 69.37 (CH₂–O), 102.50 (Ar), 127.99 (CH=, Ar), 136.54 (Ar), 143.52 (Ar, C₆₀), 156.28 (Ar–O). Anal. Calcd. for C₂₄₀H₁₅₂NaO₈: C 90.45, H 4.81%; found C 90.47, H 4.81%.

Supramolecular complex 20

Brown powder 90 mg (45% yield). UV–Vis (solid state) nm: 249, 365. IR (cm⁻¹), 3406, 3026, 2919, 2620, 1628, 1403, 1373, 1104, 1005, 967, 831, 749, 701, 573, 524; ¹³C CP-MAS NMR δ_c (ppm): 33.64 (CH, CH₂), 69.76 (CH₂–O), 128.62 (CH=, Ar), 135.55 (Ar), 143.44 (Ar, C₆₀), 161.16 (Ar–O). Anal. Calcd. for C₃₀₄H₂₀₀NaO₈: C 91.72, H 5.06%; found C 91.70, H 5.08%.

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References

- [1] Vögtle, F.; Gestermann, S.; Hesse, R.; Schwierz, H.; Windisch, B. *Prog. Polym. Sci.* **2000**, *25*, 987.
- [2] Tully, D. C.; Fréchet, J. M. J. *Chem. Commun.* **2001**, 1229.
- [3] Inoue, K. *Prog. Polym. Sci.* **2000**, *25*, 453.
- [4] Bosman, A. W.; Janssen, H. M.; Meijer, E. W. *Chem. Rev.* **1999**, *99*, 1665.
- [5] Swager, T. M. *Acc. Chem. Res.* **1998**, *31*, 201.
- [6] McQuade, D. T.; Pullen, A. E.; Swager, T.M. *Chem. Rev.* **2000**, *100*, 2537.
- [7] Sato, T.; Jiang, D.-L.; Aida, T. J. *Am. Chem. Soc.* **1999**, *121*, 10658.
- [8] Balogh, L.; de Leuze-Jallouli, A.; Dvornic, P.; Kunugi, Y.; Blumstein, A.; Tomalia, D. A. *Macromolecules* **1999**, *32*, 1036.
- [9] Schenning, A. P. H. J.; Peeters, E.; Mijer, E. W. J. *Am. Chem. Soc.* **2000**, *122*, 4489.
- [10] Swallena, S. F.; Kopelmana, R.; Mooreb, J. S.; Devadossb, C. *J. Mol. Struct.* **1999**, *485*, 585.
- [11] García, M. A.; Klimova, E.; Klimova, T.; Flores, P. B.; Romero, A. M.; Hernandez, O. S.; Martínez, G. M. *Fullerenes Nanot. Carbon Nanostruct.* **2005**, *13*, 171.
- [12] Hawker, C. J.; Fréchet, J. M. J. *Am. Chem. Soc.* **1990**, *112*, 7638.